Macromolecules are individual large polymer molecules.

A functional group is a chemically reactive atom or group of atoms that imparts characteristic properties to organic compounds containing that group.

For example: The carbonyl group C = O is the functional group for ketones.

Free radicals are very reactive atoms or molecules which have unpaired electrons.

The bond between the two coygen atoms breaks first as it is not as stable as the bond between the O-H atoms,

Polylactic acid PLA

H

0

Condensation (esterification) of lactic acid to form lactide.

O

Direct or self-polymerization is not a good option as too many water molecules are liberated,

Lactic acid is 2-hydroxypropanoic acid

Lactose is the predominant sugar in milk and enzymes turn it into lactic acid.

When a chain of lactic acid undergoes self-polymerisation, too much water is produced by the esterification (elimination) reactions.

0

It is far more efficient to place two monomers next to each other with one upside down so that the carboxylic and hydroxy functional groups are next to each other. An elimination reaction occurs as water is eliminated to produce lactide.

No initiator is needed. The reaction is slower because two steps are needed. Lactide must first form and then be polymerised. No double bonds are needed as the acid and alcohol functional groups react by eliminating water.

Holt Chemistry Carbon And Organic Compounds Answers

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