

ACID-BASE CONCEPTS

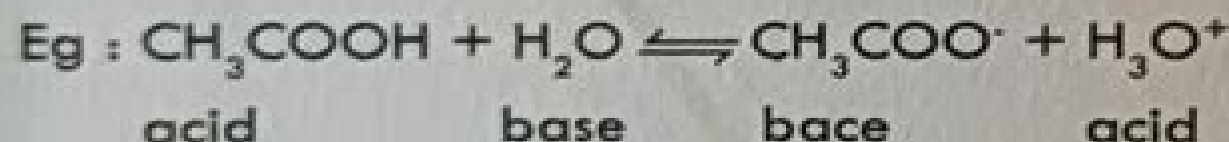
01 ARRHENIUS CONCEPT

Acids are H^+ ions donors and bases are OH^- ion donors.



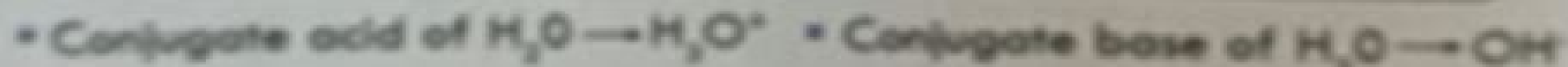
02 BRONSTED - LOWRY CONCEPT

Acids are protons (H^+ ion) donors and bases are proton acceptors



base + $H^+ \rightarrow$ Conjugate acid

acid - $H^+ \rightarrow$ Conjugate base



03 LEWIS CONCEPT

Bases are lone pair e^- donors & acids are lone pair e^- acceptors.

- Lewis acid - Substances having Vacant space in their valence shell to accomodate lone pair of e^-



- Lewis base - Substances having lone pair of e^- in their valence shell



Holt Chemistry Acid And Base Concept Review

**Michael R. Pinsky, Laurent
Brochard, Göran
Hedenstierna, Massimo Antonelli**



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The School Science Review, 2003

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