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web write the oxidation and reduction reactions for each redox reaction the first one is done for you $23 \text{ Fe} + \text{Co}^{2+} \rightarrow \text{Fe}^{2+} + \text{Co}$
 oxidation $\text{Co} + \text{Fe}^{2+} \rightarrow \text{Co}^{2+} + \text{Fe}$ reduction $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$ $24 \text{ 3 Ag} + \text{Ni}^{3+} \rightarrow 3 \text{ Ag} + \text{Ni}$ oxidation reduction $25 \text{ Cu} + \text{Pb}^{2+} \rightarrow \text{Cu}^{2+} + \text{Pb}$ oxidation reduction 26 O_2

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 $5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + \text{I}_2$ $\text{O}_5 + 5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + \text{I}_2$

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