

COVALENT BONDING

Name _____

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

*H wants 2 electrons
all other atoms 8*



Note that hydrogen is content with 2, not 8, electrons.

electrons in the middle share

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

wanted available share

1. $\text{H} + \text{H} (\text{H}_2)$ W - 4 A - 2 S - 2	<i>wanted 2 + 2</i> <i>available</i> $\text{H} \cdot \cdot \text{H}$ $\text{H} \cdot \quad \cdot \text{H}$	$\text{H} - \text{H}$ <i>single bond = share 2 e⁻</i>
2. $\text{F} + \text{F} (\text{F}_2)$ W - 8 + 8 = 16 A - 7 + 7 = 14 S - 2	$\cdot \ddot{\text{F}} \cdot \cdot \ddot{\text{F}} \cdot$ <i>2 e⁻</i>	$\cdot \ddot{\text{F}} - \ddot{\text{F}} \cdot$ <i>single bond 2 e⁻</i>
3. $\text{O} + \text{O} (\text{O}_2)$ W - 8 + 8 = 16 A - 6 + 6 = 12 S - 4	$\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{O}} \cdot$ <i>4 e⁻</i>	$\cdot \ddot{\text{O}} = \ddot{\text{O}} \cdot$ <i>4 double bond</i>
4. $\text{N} + \text{N} (\text{N}_2)$ W - 8 + 8 = 16 A - 5 + 5 = 10 S - 6	$\cdot \ddot{\text{N}} \cdot \cdot \ddot{\text{N}} \cdot$	$\cdot \ddot{\text{N}} \equiv \ddot{\text{N}} \cdot$
5. $\text{C} + \text{O} (\text{CO}_2)$ W - 8 + 8 + 8 = 24 A - 4 + 6 + 6 = 16 S - 8	$\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{C}} \cdot \cdot \ddot{\text{O}} \cdot$	$\cdot \ddot{\text{O}} = \text{C} = \ddot{\text{O}} \cdot$
6. $\text{H} + \text{O} (\text{H}_2\text{O})$ W - 2 + 2 + 8 = 12 A - 1 + 1 + 6 = 8 S - 4	$\text{H} \cdot \cdot \ddot{\text{O}} \cdot \cdot \text{H}$	$\text{H} - \ddot{\text{O}} - \text{H}$

Holt Chemistry Covalent Review Answers

BM King



Holt Chemistry Covalent Review Answers:

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Table of Contents Holt Chemistry Covalent Review Answers

1. Understanding the eBook Holt Chemistry Covalent Review Answers
 - The Rise of Digital Reading Holt Chemistry Covalent Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Covalent Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Covalent Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Covalent Review Answers
 - Personalized Recommendations

- Holt Chemistry Covalent Review Answers User Reviews and Ratings
- Holt Chemistry Covalent Review Answers and Bestseller Lists
- 5. Accessing Holt Chemistry Covalent Review Answers Free and Paid eBooks
 - Holt Chemistry Covalent Review Answers Public Domain eBooks
 - Holt Chemistry Covalent Review Answers eBook Subscription Services
 - Holt Chemistry Covalent Review Answers Budget-Friendly Options
- 6. Navigating Holt Chemistry Covalent Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Chemistry Covalent Review Answers Compatibility with Devices
 - Holt Chemistry Covalent Review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Chemistry Covalent Review Answers
 - Highlighting and Note-Taking Holt Chemistry Covalent Review Answers
 - Interactive Elements Holt Chemistry Covalent Review Answers
- 8. Staying Engaged with Holt Chemistry Covalent Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Covalent Review Answers
- 9. Balancing eBooks and Physical Books Holt Chemistry Covalent Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Chemistry Covalent Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Chemistry Covalent Review Answers
 - Setting Reading Goals Holt Chemistry Covalent Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Chemistry Covalent Review Answers
 - Fact-Checking eBook Content of Holt Chemistry Covalent Review Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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