

NUCLEAR CHEMISTRY WORKSHEET

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Most nuclear reactions involve changes in the number of protons and/or neutrons in the nucleus of an atom. These changes are called *transmutations* because an atom of one element is changed into an atom of a different element. During these changes, nuclear particles may be absorbed and/or emitted along with a release of energy.

There are four types of nuclear reactions:

- 1) *natural radioactive decay*: transmutations occur by natural radioactivity (the ability of a nucleus to emit a nuclear particle and energy without external stimulation);
- 2) *artificial transmutation*: transmutations occur by nuclear disintegration caused by external stimulation as a scientist bombards a nucleus with a particle (the addition of another nuclear particle makes the nucleus unstable);
- 3) *fission*: certain nuclei having a large mass are bombarded with special particles that cause the nuclei to split into two nuclei each having a smaller mass;
- 4) *fusion*: nuclei of light elements are combined to form heavier nuclei.

Nuclear reactions are written using symbols in the A_ZX notation. The symbols for the major nuclear particles involved in nuclear reactions are given below.

Alpha particle	${}^4_2\text{He}$ (an alpha particle is a helium nucleus)
Proton	${}^1_1\text{H}$ (the most common hydrogen nucleus is a proton)
Neutron	${}^1_0\text{n}$
Electron	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$ (also called a beta particle)
Positron	${}^0_1\text{e}$ or ${}^0_1\beta$
Gamma ray	${}^0_0\gamma$

The table below contains information about nuclear reactions involving the emission of radiation.

Type of Reaction	NUCLEAR CHANGES		
	Radiation (Particle)	Effect on the Atomic Number	Effect on the Atomic Mass
Alpha emission (α)	${}^4_2\text{He}$	decrease by 2	decrease by 4
Beta emission (β)	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$	increase by 1	no change
Positron emission (β^+)	${}^0_1\text{e}$ or ${}^0_1\beta$	decrease by 1	no change
Gamma emission	${}^0_0\gamma$	no change	no change
Electron capture (EC)	${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$	decrease by 1	no change

Holt Chemistry Concept Review Answers Nuclear Chemistry

Camilla Rothe



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Table of Contents Holt Chemistry Concept Review Answers Nuclear Chemistry

1. Understanding the eBook Holt Chemistry Concept Review Answers Nuclear Chemistry
 - The Rise of Digital Reading Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Concept Review Answers Nuclear Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Personalized Recommendations
 - Holt Chemistry Concept Review Answers Nuclear Chemistry User Reviews and Ratings
 - Holt Chemistry Concept Review Answers Nuclear Chemistry and Bestseller Lists
5. Accessing Holt Chemistry Concept Review Answers Nuclear Chemistry Free and Paid eBooks

- Holt Chemistry Concept Review Answers Nuclear Chemistry Public Domain eBooks
- Holt Chemistry Concept Review Answers Nuclear Chemistry eBook Subscription Services
- Holt Chemistry Concept Review Answers Nuclear Chemistry Budget-Friendly Options
- 6. Navigating Holt Chemistry Concept Review Answers Nuclear Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Chemistry Concept Review Answers Nuclear Chemistry Compatibility with Devices
 - Holt Chemistry Concept Review Answers Nuclear Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Highlighting and Note-Taking Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Interactive Elements Holt Chemistry Concept Review Answers Nuclear Chemistry
- 8. Staying Engaged with Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Concept Review Answers Nuclear Chemistry
- 9. Balancing eBooks and Physical Books Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Chemistry Concept Review Answers Nuclear Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Setting Reading Goals Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Fact-Checking eBook Content of Holt Chemistry Concept Review Answers Nuclear Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

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- Interactive and Gamified eBooks

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