

Lab 4 – How Much Citric Acid is in Lemon Juice?

Pre-Lab Questions:

First, inevitably, your teacher always gives you a problem relating acid to its conjugate.

- 1.1 What is the chemical formula of the acid in lemon juice? What is the name of the chemical containing water?
- The name of the acid in lemon juice is citric acid. The name of the acid in baking soda is sodium hydrogen carbonate or sodium bicarbonate.
- 1.2 Write a balanced net ionic equation for citric acid reacting with water. What chemical is responsible for the gas production during the reaction between baking soda and citric acid?
- Sodium hydrogen carbonate is the baking soda. It is responsible for the gas production.
- 1.3 At what point during the laboratory experiment do you add baking soda to citric acid solution? Where will the nitrogen bubbles appear? Where does this happen?
- When adding baking soda to citric acid solution, gas production is also noticeable. This is the indication that the reaction is taking place.
- 1.4 Write a balanced net ionic equation for the reaction between citric acid and the amount of baking soda.
- In order to find the volume amount of citric acid required for the reaction, the balanced chemical equation must be known. $2\text{NaHCO}_3 + \text{H}_3\text{C}_6\text{H}_5\text{O}_7 \rightarrow 2\text{NaC}_6\text{H}_5\text{O}_7 + \text{H}_2\text{O} + \text{CO}_2$. From the balanced equation, you know that 2 moles of sodium hydrogen carbonate react with 1 mole of citric acid for the reaction. Therefore, 0.74 mole baking soda is 1 mole citric acid. Therefore, 0.74 mole of citric acid.
- 1.5 Write a balanced net ionic equation for the reaction between citric acid and the amount of baking soda.
- In order to find the volume amount of citric acid required for the reaction, the balanced chemical equation must be known. $2\text{NaHCO}_3 + \text{H}_3\text{C}_6\text{H}_5\text{O}_7 \rightarrow 2\text{NaC}_6\text{H}_5\text{O}_7 + \text{H}_2\text{O} + \text{CO}_2$. From the balanced equation, you know that 2 moles of sodium carbonate react with 1 mole of citric acid for the reaction. Therefore, 0.74 mole baking soda is 1 mole citric acid. Therefore, 0.74 mole of citric acid.

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