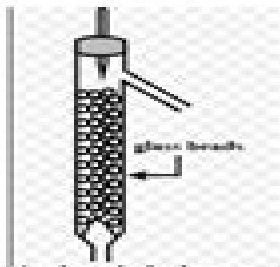


|                                                                                                                                     |                                                                                                                                                                                                                                                                                                                                                                                                                                |                                                                                                                                                                                                                   |                  |
|-------------------------------------------------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|------------------|
| Suggest why chloroethanoic acid is stronger than ethanoic acid                                                                      | (this question is already discussed a little further down.)                                                                                                                                                                                                                                                                                                                                                                    |                                                                                                                                                                                                                   |                  |
| fill in rate table                                                                                                                  | 1.00<br>0.16<br>1.35<br>I beg someone write what the values were if they remember                                                                                                                                                                                                                                                                                                                                              | K=250<br>Yes i agree - needs to be 1.4 as all values were to two sig figs<br>Weren't we they to 3 dp though?<br>The rate of the first one was 1.00 so I'm guessing 3sf of 2dp                                     | 3                |
| empirical formula                                                                                                                   | C6H10O3S2                                                                                                                                                                                                                                                                                                                                                                                                                      | Percentages dudbt add up to 100 fuck me<br>Didn't even see the sulphur, fml same same i thought i meant i was in the fuel area. YAY i got this!!                                                                  | 3                |
| the labeling distillation                                                                                                           | for the tube it was fractionating column or glass beads<br>the bottom of the condenser was water in<br>and top was water out                                                                                                                                                                                                                                                                                                   |  <p>reject anti bumping granules<br/>might accept silica BEADS</p>                                                             | 3                |
| Draw a line for the reaction with chloroethanoic acid<br><br>Suggest why chloroethanoic acid is a stronger acid than ethanoic acid. | <p>Was the line of chloroethanoic acid to the left of the first line, but levelled off at the same point?<br/>Yes but it levelled out before, same height though. Ah ok.</p> <p>Dissociates more<br/>Chlorine is more electronegative than hydrogen so will withdraw electron density from the carboxyl group meaning that the O-H bond breaks more easily dissociating into H<sup>+</sup> making the solution more acidic</p> | Cl is electron withdrawing. it polarises the bond, weakens the O-H bond which releases more H <sup>+</sup> than ethanoic acid<br>Neg inductive effect?<br>Chloroethanoic acid forms a more stable conjugate base? | 3                |
| NMR Question?                                                                                                                       | <p>Use the integration values and the ratios to identify the structure at two peaks.</p> <p>Draw this structure.</p>                                                                                                                                                                                                                                                                                                           | Was there a typo though? Regarding what?<br>No dont think so, it was just because 2 of the peaks were the same environment almost so people got thrown off                                                        | 5<br>2<br>1<br>2 |

# F335 June 2013 Unofficial Mark Scheme

**M Lipman**



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### **Table of Contents F335 June 2013 Unofficial Mark Scheme**

1. Understanding the eBook F335 June 2013 Unofficial Mark Scheme
  - The Rise of Digital Reading F335 June 2013 Unofficial Mark Scheme
  - Advantages of eBooks Over Traditional Books
2. Identifying F335 June 2013 Unofficial Mark Scheme
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an F335 June 2013 Unofficial Mark Scheme
  - User-Friendly Interface
4. Exploring eBook Recommendations from F335 June 2013 Unofficial Mark Scheme
  - Personalized Recommendations

- F335 June 2013 Unofficial Mark Scheme User Reviews and Ratings
- F335 June 2013 Unofficial Mark Scheme and Bestseller Lists
- 5. Accessing F335 June 2013 Unofficial Mark Scheme Free and Paid eBooks
  - F335 June 2013 Unofficial Mark Scheme Public Domain eBooks
  - F335 June 2013 Unofficial Mark Scheme eBook Subscription Services
  - F335 June 2013 Unofficial Mark Scheme Budget-Friendly Options
- 6. Navigating F335 June 2013 Unofficial Mark Scheme eBook Formats
  - ePub, PDF, MOBI, and More
  - F335 June 2013 Unofficial Mark Scheme Compatibility with Devices
  - F335 June 2013 Unofficial Mark Scheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of F335 June 2013 Unofficial Mark Scheme
  - Highlighting and Note-Taking F335 June 2013 Unofficial Mark Scheme
  - Interactive Elements F335 June 2013 Unofficial Mark Scheme
- 8. Staying Engaged with F335 June 2013 Unofficial Mark Scheme
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers F335 June 2013 Unofficial Mark Scheme
- 9. Balancing eBooks and Physical Books F335 June 2013 Unofficial Mark Scheme
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection F335 June 2013 Unofficial Mark Scheme
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine F335 June 2013 Unofficial Mark Scheme
  - Setting Reading Goals F335 June 2013 Unofficial Mark Scheme
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of F335 June 2013 Unofficial Mark Scheme
  - Fact-Checking eBook Content of F335 June 2013 Unofficial Mark Scheme

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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