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 $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg(NO}_3)_2 + \text{H}_2 + \text{O}_2$
 automatically balanced by balancing nitrogen

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 g H_2O l b $\text{Mg}^{2+} + \text{CO}_3^{2-} + 2\text{H}^+ \rightarrow \text{Mg}^{2+} + 2\text{NO}_3^- + \text{H}_2\text{O}$ write the balanced equation when nitric acid reacts with calcium hydroxide to produce water and calcium nitrate

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