

A-Level Chemistry - Year 2

OCR Chemistry A - Paper 1 June - 2019

16 Sir Humphrey Davy discovered several elements including sodium, potassium, magnesium, calcium and strontium.

(a) Explain which block in the Periodic Table sodium and magnesium belong to.

(1)

(b) A sample of magnesium, $A_r = 24.305$, is found to consist of three isotopes. The accurate relative isotopic masses and % abundances of two of the isotopes are shown in the table.

Isotope	Relative isotopic mass	% abundance
^{24}Mg	23.985	78.99%
^{26}Mg	24.986	10.00%

Determine the relative isotopic mass of the third isotope of magnesium in the sample.

Give your answer to 5 significant figures.

relative isotopic mass = _____ (2)

Question - 16

OCR

Official Cambridge and CIE

Tuesday 4 June 2019 – Afternoon

A Level Chemistry A

443201 Periodic table, elements and physical chemistry

Time allowed: 1 hour 15 minutes

We need from:

1. An A-level Chemistry A (or A-level Chemistry A and A-level General Chemistry)

We also need:

1. A minimum of 100 pages of paper



Please write clearly in black ink. Do not write in the brackets.

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Candidate number

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INSTRUCTIONS

- 1. Use black ink. You may use an HB pencil for graphs and diagrams.
- 2. Answer all the questions.
- 3. Where appropriate, your answers should be supported with working. Marks may be given for a correct method even if the answer is incorrect.
- 4. Write your answer to each question in the space provided. If additional space is required, use the back paper(s) at the end of this booklet. The question numbers must be clearly shown.

INFORMATION

- 1. The total mark for this paper is 100.
- 2. The marks for each question are shown in brackets ().
- 3. Quality of extended responses will be assessed in questions marked with an asterisk (*).
- 4. This document consists of 10 pages.

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DP Hallahan



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Table of Contents F321 Chemistry June 2014 Paper

1. Understanding the eBook F321 Chemistry June 2014 Paper
 - The Rise of Digital Reading F321 Chemistry June 2014 Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying F321 Chemistry June 2014 Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an F321 Chemistry June 2014 Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from F321 Chemistry June 2014 Paper
 - Personalized Recommendations

- F321 Chemistry June 2014 Paper User Reviews and Ratings
- F321 Chemistry June 2014 Paper and Bestseller Lists
- 5. Accessing F321 Chemistry June 2014 Paper Free and Paid eBooks
 - F321 Chemistry June 2014 Paper Public Domain eBooks
 - F321 Chemistry June 2014 Paper eBook Subscription Services
 - F321 Chemistry June 2014 Paper Budget-Friendly Options
- 6. Navigating F321 Chemistry June 2014 Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - F321 Chemistry June 2014 Paper Compatibility with Devices
 - F321 Chemistry June 2014 Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of F321 Chemistry June 2014 Paper
 - Highlighting and Note-Taking F321 Chemistry June 2014 Paper
 - Interactive Elements F321 Chemistry June 2014 Paper
- 8. Staying Engaged with F321 Chemistry June 2014 Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers F321 Chemistry June 2014 Paper
- 9. Balancing eBooks and Physical Books F321 Chemistry June 2014 Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection F321 Chemistry June 2014 Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine F321 Chemistry June 2014 Paper
 - Setting Reading Goals F321 Chemistry June 2014 Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of F321 Chemistry June 2014 Paper
 - Fact-Checking eBook Content of F321 Chemistry June 2014 Paper

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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