

Name : \_\_\_\_\_ Date : \_\_\_\_\_

# Acids and Bases Strength

Write True (T) or False (F).

- i. Stronger acids have weaker conjugate bases and weaker acids have stronger conjugate bases. \_\_\_\_\_
- ii. Stronger bases have stronger conjugate acids and weaker bases have weaker conjugate acids. \_\_\_\_\_

Write equations that show how the following reactions occur in water.

- i.  $F^-$  acts as a base. \_\_\_\_\_
- ii.  $HNO_2$  acts as an acid. \_\_\_\_\_
- iii.  $Fe(H_2O)_6^{3+}$  acts as an acid. \_\_\_\_\_
- iv.  $HCO_3^-$  acts as a base. \_\_\_\_\_
- v.  $HCO_3^-$  acts as an acid. \_\_\_\_\_
- vi.  $Al(H_2O)_5(OH)^{2+}$  acts as a base. \_\_\_\_\_

What volume of 0.45 M LiOH would be needed to neutralize 60.0 mL of 0.15 M HI?

Calculate the pH values of the following solutions of strong acids and bases.

- i. 0.001 M HCl
- ii. 0.76 M KOH
- iii.  $2.8 \times 10^{-4}$  M  $Ba(OH)_2$

# Lecture 29 Acids And Bases I Worksheet

**ML Morrison**



## **Lecture 29 Acids And Bases I Worksheet:**

## Unveiling the Magic of Words: A Review of "**Lecture 29 Acids And Bases I Worksheet**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Lecture 29 Acids And Bases I Worksheet**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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