

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 18. It varies on the periodic table.
- _____ 19. It has no fixed periodic table.
- _____ 20. Group 1 elements.
- _____ 21. Elements that are strong oxidizing agents.
- _____ 22. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of elements.
- _____ 24. Most of the elements of the periodic table are metals.
- _____ 25. Group 1A elements except for hydrogen are metals on the left.
- _____ 26. Group 1A elements are the alkali metal metals.
- _____ 27. Group 1A elements are highly reactive elements on the left.
- _____ 28. Group 1A elements are very reactive elements on the left.
- _____ 29. Alkali metals have properties of both metals and non-metals.

Hydrocarbons Study Guide Answers For Content Mastery

Robert J. McCunney, Paul P. Rountree



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