

Concept Review

Section: What Is Matter?

1. **Classify** the following as a homogeneous, *O*, or a heterogeneous, *E*, mixture.

_____ **a.** a pail of sand and water

_____ **b.** air

_____ **c.** human blood

_____ **d.** banana split

_____ **e.** chocolate syrup

_____ **f.** sea water

2. **Compare and contrast** atoms and molecules.

3. **Explain** the difference between a pure substance and a homogeneous mixture.

4. **Classify** each of the following as an element or a compound.

_____ **a.** benzene, C_6H_6

_____ **b.** aluminum, Al

_____ **c.** aspirin, $C_9H_8O_4$

_____ **d.** titanium, Ti

_____ **e.** acetylene, C_2H_2

_____ **f.** zinc, Zn

5. **Explain** why elements and compounds are pure substances.

6. **Determine** which of the following are pure substances and which are mixtures.

_____ **a.** salt water

_____ **b.** isopropyl alcohol, C_3H_8O

_____ **c.** mercury, Hg

_____ **d.** ammonia, NH_3

_____ **e.** an egg yolk

_____ **f.** honey

Holt Chemistry Section 2 Answers

Roman Wölfel

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