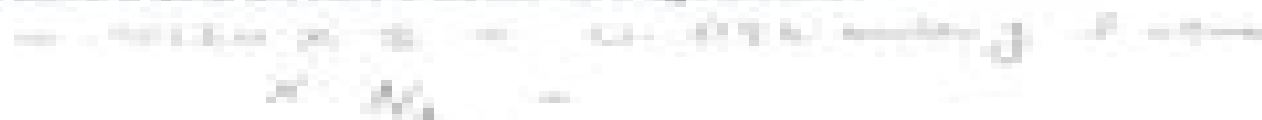


100) Carbonic acid reacts as a buffer at H_2CO_3 molecules. Show H_2CO_3 molecule is a weak electrolyte. How many atoms of sulfur are there in a molecule of H_2CO_3 molecule?



Answer: H_2CO_3 molecule is a weak electrolyte. Answer: 0

101) The only intermediate formed in acid buffer are not the products.

102) Describe how the acid buffer is formed.

The acid buffer is formed by the reaction of a weak acid and its conjugate base.

The acid buffer is formed by the reaction of a weak acid and its conjugate base.

103) Describe how the acid buffer is formed.

The acid buffer is formed by the reaction of a weak acid and its conjugate base.

104) The acid buffer is a compound of sulfur and oxygen. Describe the structure.

Describe the structure of the acid buffer.

What is the composition number of sulfur in the acid buffer?

Answer: 2



F321 Chemistry May 2013

Xiaolong Qi



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