# CHEMICAL EQUATIONS AND WORKSHEET

# 1.Using the following equation:

$$Pb(SO_4)_2 + 4 LiNO_3 \rightarrow Pb(NO_3)_4 + 2 Li_2SO_4$$

How much (in grams) lithium nitrate will be needed to make 250 grams of lithium sulfate, assuming you have enough lead (IV) sulfate to react?

#### 2. Using the following equation:

$$Fe_2O_3 + 3H_2 \rightarrow 2Fe + 3H_2O$$
.

Calculate how many grams of iron can be made from 16.5 grams of  $Fe_2O_3$ .

#### 3. Using the following equation:

$$5 \text{ KNO}_2 + 2 \text{ KMnO}_4 + 3 \text{ H}_2 \text{SO}_4 \rightarrow 5 \text{ KNO}_3 + 2 \text{ MnSO}_4 + \text{K}_2 \text{SO}_4 + 3 \text{ H}_2 \text{O}_4$$

How many moles and grams of  $KMnO_4$  are needed for this reaction on 11.4 grams of  $KNO_4$ ?

# 4. Using the following equation:

$$2 I_2 + KIO_3 + 6 HCI \rightarrow 5 ICI + KCI + 3 H_2O$$

Calculate how many grams of iodine are needed to prepare 28.6 grams of ICI by this reaction.

# **Holt Stoichiometry Worksheet**

Holt, Rinehart and Winston Staff, Holt, Rinehart, and Winston, inc

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