

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question, using the letters in each "box" to identify the required element(s).

The image shows a crossword puzzle grid with letters placed in specific cells. The letters are as follows:

- Row 1:** A (1,1), B (1,13)
- Row 2:** C (2,1), D (2,10), E (2,12)
- Row 3:** F (3,2), G (3,7), H (3,10), K (3,13)
- Row 4:** L (4,2), I (4,12)
- Row 5:** J (5,1), Z (5,3), M (5,7), N (5,10), O (5,13), P (5,14)
- Row 6:** Q (6,1), R (6,3), S (6,7), T (6,8), U (6,9)
- Row 7:** V (7,1), W (7,10)
- Row 8:** X (8,1), Y (8,10)

Arrows point from the letter 'R' (6,3) to the starting cells of two words in the bottom row: 'V' (7,1) and 'X' (8,1).

List elements in alphabetical order when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "E"?
- _____ 4) Which letter labels the halogen element with the lowest atomic number?
- _____ 5) Which of these elements is the least reactive? **A)** "D" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- _____ 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

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RD Boyd



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