

Chapter 10 Test Form A

1. B
2. A
3. A
4. B
5. D
6. B
7. B
8. A
9. A
10. A
11. B
12. D
13. A
14. C
15. No, "heat" and "cold" do not flow between objects. Energy transferred between objects changes the temperature of the object.
16. An object at high temperature has higher-energy particles and an object at low temperature has lower-energy particles. Energy is transferred as heat from the higher-energy particles to the lower-energy particles through collisions of the higher with the lower energy particles.
17. The ice begins to melt and change into water.
18. Thermal conduction is the process by which energy is transferred as heat through a material between two points at different temperatures. The atoms of the hotter substance lose energy and the atoms of the cooler substance gain energy.
19. Hypothermia is the transfer of too much energy from the human body to the surrounding air so that body temperature falls below normal levels.
20. Covering the body would protect it from direct sunlight and prevent excessive loss of body water through evaporation.

Chapter 10 Test Form B

1. B
2. D
3. C
4. A
5. C
6. A
7. A
8. C
9. B
10. A
11. B
12. B
13. D
14. A
15. C
16. When energy is added to the gas, the kinetic energy of the molecules increases. The temperature increases because temperature is proportional to kinetic energy of the molecules.
17. The molecules of water have a higher kinetic energy, so there are more particle collisions per time per mass in the water than in the hand. Energy is transferred as heat to the outer layers of skin on the hand, which has fewer molecular collisions per time per mass.
18. The temperature stops rising, and the water turns into steam.
19. Air is an extremely poor thermal conductor. A thin layer of air near the skin provides a barrier from energy transfer.
20. Covering the body would protect it from direct sunlight and prevent excessive loss of body water through evaporation.

Holt Physics Chapter Test A Heat

Terry C. Jones



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