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**MULTIPLE CHOICE.** Choose the one alternative that best completes the statement or answers the question.

- 1) Which of the following statements about the atom  $^{12}_6\text{C}$  is FALSE? 1) \_\_\_\_\_
- A) It has 12 neutrons in its nucleus.
  - B) It has 6 protons in its nucleus.
  - C) Its atomic weight is 12.
  - D) It has 6 electrons orbiting the nucleus.
  - E) Its atomic number is 6.

Table 2.1

O C H

- 2) Using the information in Table 2.1, calculate the molecular weight of ethanol,  $\text{C}_2\text{H}_5\text{OH}$ . 2) \_\_\_\_\_
- A) 33
  - B) 96
  - C) 46
  - D) 34
  - E) The answer cannot be determined.
- 3) Antacids neutralize acid by the following reaction. Identify the salt in the following equation: 3) \_\_\_\_\_
- $$\text{Mg}(\text{OH})_2 + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$$
- A)  $\text{H}_2\text{O}$
  - B)  $\text{MgCl}_2$
  - C)  $\text{HCl}$
  - D)  $\text{Mg}(\text{OH})_2$
  - E) None of the answers is correct.
- 4) Which of the following statements is FALSE? 4) \_\_\_\_\_
- A) Water freezes from the top down.
  - B) Salts readily dissolve in water.
  - C) Water molecules are formed by hydrolysis.
  - D) Water is a part of a dehydration synthesis reaction.
  - E) Water is a polar molecule.
- 5) Which of the following is the type of bond holding  $\text{K}^+$  and  $\text{I}^-$  ions in KI? 5) \_\_\_\_\_
- A) hydrogen bond
  - B) ionic bond
  - C) covalent bond
- 6) Which of the following is the type of bond between molecules of water in a beaker of water? 6) \_\_\_\_\_
- A) hydrogen bond
  - B) covalent bond
  - C) ionic bond

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**R Barnett**



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