

3. A student wants to remove an acid impurity from an organic liquid.  
What should the student do?

- A. Add  $\text{Na}_2\text{CO}_3(\text{aq})$
- B. Reflux the mixture
- C. Add  $\text{Br}_2$
- D. Add  $\text{MgSO}_4$

Your answer

[1]

4. For which purpose is distillation used?
- A. to allow a liquid to boil without the loss of vapour
  - B. to purify a liquid product
  - C. to remove an involatile impurity
  - D. to allow further reaction without the loss of product

Your answer

[1]

5. The reaction below produces a chloroalkane.



Which row shows the steps to purify the liquid product in the correct order?

<b>A</b>	Use a separating funnel	Distil	Dry	Remove unreacted $\text{HCl}$
<b>B</b>	Remove unreacted $\text{HCl}$	Use a separating funnel	Dry	Distil
<b>C</b>	Remove unreacted $\text{HCl}$	Use a separating funnel	Distil	Dry
<b>D</b>	Use a separating funnel	Remove unreacted $\text{HCl}$	Distil	Dry

Your answer

[1]

# May 2014 Chemistry Ocr Mark Scheme

**TD Snyder**



**May 2014 Chemistry Ocr Mark Scheme:**

## Unveiling the Magic of Words: A Review of "**May 2014 Chemistry Ocr Mark Scheme**"

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