(Question 1 continued)

(d) The cross-sectional area of the tube is 1.3 × 10⁻³ m² and the temperature of air is 300 K. Estimate the number of moles of air in the tube.

[2]

[2]

(e) The equation in (b) may be used to predict the pressure of the air at extremely large values of ¹/_H. Suggest why this will be an unreliable estimate of the pressure.

Very large 1/H means small volume. Small volume means high cressure.

In ideal gas law, it stated that molecules are not close to each other and hence the forces between them are negligible.

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- 1. Understanding the eBook May 2013 Ib Paper 3 Markscheme Physics
 - The Rise of Digital Reading May 2013 Ib Paper 3 Markscheme Physics
 - Advantages of eBooks Over Traditional Books
- 2. Identifying May 2013 Ib Paper 3 Markscheme Physics
 - Exploring Different Genres
 - $\circ\,$ Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an May 2013 Ib Paper 3 Markscheme Physics
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from May 2013 Ib Paper 3 Markscheme Physics

- Personalized Recommendations
- May 2013 Ib Paper 3 Markscheme Physics User Reviews and Ratings
- May 2013 Ib Paper 3 Markscheme Physics and Bestseller Lists
- 5. Accessing May 2013 Ib Paper 3 Markscheme Physics Free and Paid eBooks
 - May 2013 Ib Paper 3 Markscheme Physics Public Domain eBooks
 - May 2013 Ib Paper 3 Markscheme Physics eBook Subscription Services
 - May 2013 Ib Paper 3 Markscheme Physics Budget-Friendly Options
- 6. Navigating May 2013 Ib Paper 3 Markscheme Physics eBook Formats
 - o ePub, PDF, MOBI, and More
 - May 2013 Ib Paper 3 Markscheme Physics Compatibility with Devices
 - May 2013 Ib Paper 3 Markscheme Physics Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of May 2013 Ib Paper 3 Markscheme Physics
 - Highlighting and Note-Taking May 2013 Ib Paper 3 Markscheme Physics
 - Interactive Elements May 2013 Ib Paper 3 Markscheme Physics
- 8. Staying Engaged with May 2013 Ib Paper 3 Markscheme Physics
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers May 2013 Ib Paper 3 Markscheme Physics
- 9. Balancing eBooks and Physical Books May 2013 Ib Paper 3 Markscheme Physics
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection May 2013 Ib Paper 3 Markscheme Physics
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine May 2013 Ib Paper 3 Markscheme Physics
 - Setting Reading Goals May 2013 Ib Paper 3 Markscheme Physics
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of May 2013 Ib Paper 3 Markscheme Physics

- Fact-Checking eBook Content of May 2013 Ib Paper 3 Markscheme Physics
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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