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Manual For 64

**United States. Department of the
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Manual For 64:

Aviation Unit and Intermediate Troubleshooting Manual for Army Model AH-64A Helicopter (NSN 1520-01-106-9519)
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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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