

Section 1: Chemistry

SECTION 1

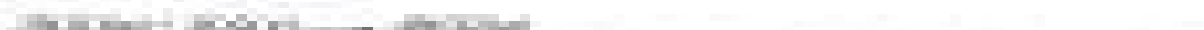
Answer each question on the following questions in the space provided.

1. ☐ a. The following are chemical equations representing the reaction of hydrogen gas with oxygen gas.
(a) $2H_2 + O_2 \rightarrow 2H_2O$
(b) $2H_2 + O_2 \rightarrow 2H_2O$
(c) $2H_2 + O_2 \rightarrow 2H_2O$
(d) $2H_2 + O_2 \rightarrow 2H_2O$
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(f) $2H_2 + O_2 \rightarrow 2H_2O$
(g) $2H_2 + O_2 \rightarrow 2H_2O$
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(q) $2H_2 + O_2 \rightarrow 2H_2O$
(r) $2H_2 + O_2 \rightarrow 2H_2O$
(s) $2H_2 + O_2 \rightarrow 2H_2O$
(t) $2H_2 + O_2 \rightarrow 2H_2O$
(u) $2H_2 + O_2 \rightarrow 2H_2O$
(v) $2H_2 + O_2 \rightarrow 2H_2O$
(w) $2H_2 + O_2 \rightarrow 2H_2O$
(x) $2H_2 + O_2 \rightarrow 2H_2O$
(y) $2H_2 + O_2 \rightarrow 2H_2O$
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3. ☐ c. The following are chemical equations representing the reaction of hydrogen gas with oxygen gas.
(a) $2H_2 + O_2 \rightarrow 2H_2O$
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(x) $2H_2 + O_2 \rightarrow 2H_2O$
(y) $2H_2 + O_2 \rightarrow 2H_2O$
(z) $2H_2 + O_2 \rightarrow 2H_2O$

4. Write the balanced chemical equation for the following reaction:



a. Balance the equation:



b. Write the balanced chemical equation for the reaction:



c. Write the balanced chemical equation for the reaction:



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L Cohen



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