

1 Temperature**BEFORE YOU READ**

After you read this section, you should be able to answer these questions:

- How are temperature and kinetic energy related?
- How is temperature measured?
- What is thermal expansion?

National Science
Education Standards
PS 1a, 1b

What Is Temperature?

You may think of temperature as how hot or cold something is. But using the words *hot* and *cold* can be confusing. Pretend that you are outside on a hot day. If you step onto a porch with a fan blowing, you might think it feels cool. Then, your friend enters the porch from an air-conditioned house. She thinks the porch is warm!

Using the words *cool* and *warm* to describe the porch is confusing. Measuring the temperature on the porch tells you exactly how hot or cold it is. The **temperature** tells you the average kinetic energy of the particles in an object. **50**

TEMPERATURE AND KINETIC ENERGY

All matter is made of atoms or molecules that are always moving. The particles that are moving have kinetic energy. The faster they move, the more kinetic energy they have. Look at the figure below. The more kinetic energy the particles have, the higher the temperature is.



The gas particles on the right have higher kinetic energy than those on the left.

STUDY TIP

Describe Describe how temperature affects the average kinetic energy and thermal expansion. Give several examples of thermal expansion.

READING CHECK

1. **Identify** What is used to tell how hot or cold something is?

TAKE A LOOK

2. **Identify** Which gas shows particles at the higher temperature?

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Victor M. Corman



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