

Chapter Test B, continued

- _____ 8. In a reaction, the ions of two compounds exchange places to produce another set of two ions and compounds. This reaction is called a _____.
- synthesis reaction
 - decomposition reaction
 - single-displacement reaction
 - double-displacement reaction
- _____ 9. The use of a double arrow in a chemical equation indicates that the reaction _____.
- is reversible
 - requires heat
 - is written backward
 - has not been confirmed in the laboratory

PART II Write the correct term (or terms) in the space provided.

10. A chemical equation is solved that begins with heat of a substance in a chemical equation is called ΔH . _____
11. In a chemical equation, the symbol (l) indicates that the substance is _____.
12. The product is written for a single reactant as (aq) . _____
13. The general equation for a synthesis reaction is _____.
14. The general equation for a decomposition reaction is _____.
15. The general equation for a single-displacement reaction is _____.
16. The general equation for a double-displacement reaction is _____.

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