2.6 - Periodic Trends worksheet

Name L Short answer 1. Label each group on the periodic table. Explain what valence electrons are. 3. Explain what makes an element stable? II. Answer the questions using the symbols on the periodic table _1.What elements are in period 5? What element(s) is a 2.halogens fitransition metals. 3. lanthanides Zalkali metals 4alkaline earth metals 8.actinides Snoble gases What elements are in the sublevel 11. d 9 8 10 p 12. 0 What elements are ____13. metals. ___14, nonmetals _____ 15.metalloids. _____16. Which element is the most reactive nonmetal? 17. Which element is the most reactive metal? ___18.Which element below is the least reactive? ____19. Which element has the largest radius in period 4? 20.Which element has the largest radius in the Alkali metals family? 21. Which element has the highest ionization energy in period four? ____22.Which elements has the second largest atomic number? 23. Which element has the largest atomic mass? 24. Which element is located at period 4, group 15? 25. Which element is located at period 5, group 2? 27 Which elements will form +1 ions? 28. Which group of metals will have the same properties 29 Which group of nonmetals will have the same properties 30. On the back, define all chapter 5, section 3 vocabulary terms.

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