

# CHEMISTRY

2013

Time: 30 minutes

Max. Marks: 15

## SECTION "A" (MULTIPLE CHOICE QUESTIONS)

1. Choose the correct answer for each from the given options:

- (i) it is a trial idea:  
(a) Experiment (b) Hypothesis (c) Theory (d) Scientific Law
- (ii) It is the empirical formula of glucose:  
(a) CH<sub>2</sub>O (b) C<sub>2</sub>H<sub>4</sub>O<sub>2</sub> (c) C<sub>5</sub>H<sub>10</sub>O<sub>5</sub> (d) C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>
- (iii) It is not an isotope of oxygen:  
(a) <sup>16</sup><sub>8</sub>O (b) <sup>17</sup><sub>8</sub>O (c) <sup>18</sup><sub>8</sub>O (d) <sup>19</sup><sub>8</sub>O
- (iv) The vertical columns of the periodic table are called:  
(a) Periods (b) Series (c) Groups (d) Rows
- (v) It is an ionic compound:  
(a) H<sub>2</sub>O (b) CO<sub>2</sub> (c) NH<sub>3</sub> (d) NaCl
- (vi) Neither definite shape nor definite volume is the property of:  
(a) Solid (b) Liquid (c) Gas (d) None of them
- (vii) It is defined as the number of moles of solution dissolved in one litre of solution:  
(a) Molarity (b) Molality (c) Normality (d) Mole fraction
- (viii) It is the unit of electrical current:  
(a) volt (b) coulomb (c) faraday (d) ampere
- (ix) It is a proton-donor:  
(a) Acid (b) Base (c) Salt (d) Alkali
- (x) It is an exothermic process:  
(a) Evaporation (b) Respiration  
(c) Thermal decomposition (d) Melting
- (xi) It is an ion-exchanger resin:  
(a) Na-zeolite (b) Ca(OH)<sub>2</sub> (c) MgSO<sub>4</sub> (d) NaCl
- (xii) When carbon burns in air, this gas is produced:  
(a) SO<sub>2</sub> (b) H<sub>2</sub>S (c) NH<sub>3</sub> (d) CO<sub>2</sub>
- (xiii) It is the study of Hydrocarbons and their derivatives:  
(a) Physical Chemistry (b) Inorganic Chemistry  
(c) Organic Chemistry (d) Environmental Chemistry
- (xiv) This gas is used for respiration:  
(a) Hydrogen (b) Oxygen (c) Nitrogen (d) Chlorine
- (xv) It is the only liquid metal:  
(a) Bromine (b) Mercury (c) Gold (d) Sulphur
- (xvi) The most abundant and useful halogen is:  
(a) Fluorine (b) Chlorine (c) Bromine (d) Iodine
- (xvii) Red ink is normally prepared from:

# Grade 2013 Chemistry Paper Final

**LM Reder**

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web if our equation has two logarithms we can use a property that says that if  $\log_a m = \log_a n$  then it is true that  $m = n$  this is the one to one property of logarithmic equations one to one property of logarithmic equations for  $m > 0$   $n > 0$   $a > 0$  and  $a \neq 1$  is any real number if  $\log_a m = \log_a n$  then  $m = n$

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web voiceover let s get some practice solving some exponential equations and we have one right over here we have  $26$  to the  $9x$  plus five power equals one so pause the video and see if you can tell me what  $x$  is going to be well the key here is to realize that  $26$  to the zeroth power to the zeroth power is equal to one

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