Polyatomic Ions Chart

Formula	Name	Formula	Name
NH4.	Ammonium	CrO ₄ -2	Chromate
NH ₃	Ammonia	Cr2O7-2	Dichromate
C2H3O2"	Acetate	MnO ₄ "	Permanganate
CH₃COO	Acetate	MnO ₄ -≥	Manganate
CN	Cyanide	NO ₂	Nitrite
CO3-2	Carbonate	NO ₃	Nitrate
HCO3	Bicarbonate	OH.	Hydroxide
C2O4 2	Oxalate	PO4-3	Phosphate
CIO.	Hypochlorite	SCN"	Thiocyanate
CIO₂	Chlorite	Fe(CN)6-3	Ferricyanide
CIO3	Chlorate	5O₃ ⁻²	Sulfite
CIO ₄	Perchlorate	50 ₄ -2	Sulfate
5₂O₃-²	Thiosulfate	HSO₄	Hydrogen sulfate
BrO	Hypobromite	IO ₃	Iodate
AsO2-3	Arsenite	SeO ₄ -2	Selenate
BrO ₃	Bromate	H5O₃ ⁻	Hydrogen sulfite

Commonly Used Multivalent Metals

Symbol	Name	-ous	-ic
Fe (Ferrum)	Iron	+2	+3
Pb (Plumbum)	Lead	+2	+4
Sn (Stannum)	Tin	+2	+4
Hg (Hydrargyum)	Mercury	+1	+2
Cu (Cuprum)	Copper	+1	+2

Prefixes

Mono=one	Hexa = six	
Di = two	Hepta = seven	
Tri = three	Octa = eight	
Tetra = four	Nona = nine	
Penta = five	Deca = ten	

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G Orfield

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This will help guide comparisons of previous and future work Chapter 4 was the first attempt to include the excited electronic state 2 in the partition function of ArO as well as the first to address the different dissociation products of the ground and first electronic levels of ArO Chapter 5 reports an interesting source of memory in ICP MS that could affect mathematical corrections for polyatomic ions For future work on these topics I suggest the following experiments and investigations Clearly not an extensive list they are instead the first topics curiosity brings to mind 1 Measurement of T sub gas values when using the flow injection technique of Appendix B It was believed that there was a fundamental difference in the plasma when the auto sampler was used versus a continuous injection Is this reflected in T sub gas values 2 The work of Chapter 3 can be expanded and supplemented with more trials new cone materials i e copper stainless steel and more cone geometries Some of this equipment is already present in the laboratory others could be purchased or made 3 T sub gas values from Chapter 3 could be correlated with instrument pressures during the experiment Pressures after the skimmer cone were recorded for many days but have yet to be collated with the measured T sub gas values 4 The work in Chapter 5 could be expanded to include more metals Does the curious correlation between measured T sub gas and element boiling point persist 5 Investigate non linear correlations to T sub gas values of the MO memory in Chapter 5 Temperatures along the skimmer walls are not a linear gradient Ring deposits have been observed on the cone and photographs of the interface show light intensities shaping a sort of tailing peak along the outside skimmer wall Is there a physical property of the metals or metal oxides that would give this peak with the T sub gas values 6 Chemical state speciation of the metal deposits on the skimmers of Chapter 5 There may be a more logical correlation between Tgas and a physical property of the deposit ing chemical if all the metals do not deposit in the same form 7 A collaboration with our computational collegues would be most welcome Newer calculations for ArO and RuO would be very helpful **Investigations Into the Origins of Polyatomic Ions in Inductively Coupled Plasma-mass Spectrometry** Sally M. McIntyre, 2010 Fragmentation of Diatomic and Polyatomic Ions in the Gas Phase Christopher John Proctor, 1981 Experimental Studies of Polyatomic Ion Interactions with Clean and Adsorbate Covered Metal Surfaces Samuel B. Wainhaus, 1997 Unimolecular and Collision-induced Dissociation Study of Polyatomic Ions at High Collision Energy Xuedong Zhou, 2001 The Particulate Nature of Polyatomic Ions, 2007

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- 1. Understanding the eBook Flinn Scientific Polyatomic Ions
 - The Rise of Digital Reading Flinn Scientific Polyatomic Ions
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Flinn Scientific Polyatomic Ions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Flinn Scientific Polyatomic Ions
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Flinn Scientific Polyatomic Ions
 - Personalized Recommendations
 - Flinn Scientific Polyatomic Ions User Reviews and Ratings
 - Flinn Scientific Polyatomic Ions and Bestseller Lists
- 5. Accessing Flinn Scientific Polyatomic Ions Free and Paid eBooks
 - Flinn Scientific Polyatomic Ions Public Domain eBooks
 - Flinn Scientific Polyatomic Ions eBook Subscription Services
 - Flinn Scientific Polyatomic Ions Budget-Friendly Options

- 6. Navigating Flinn Scientific Polyatomic Ions eBook Formats
 - o ePub, PDF, MOBI, and More
 - Flinn Scientific Polyatomic Ions Compatibility with Devices
 - Flinn Scientific Polyatomic Ions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Flinn Scientific Polyatomic Ions
 - Highlighting and Note-Taking Flinn Scientific Polyatomic Ions
 - Interactive Elements Flinn Scientific Polyatomic Ions
- 8. Staying Engaged with Flinn Scientific Polyatomic Ions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - $\circ\,$ Following Authors and Publishers Flinn Scientific Polyatomic Ions
- 9. Balancing eBooks and Physical Books Flinn Scientific Polyatomic Ions
 - $\circ\,$ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Flinn Scientific Polyatomic Ions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Flinn Scientific Polyatomic Ions
 - Setting Reading Goals Flinn Scientific Polyatomic Ions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Flinn Scientific Polyatomic Ions
 - Fact-Checking eBook Content of Flinn Scientific Polyatomic Ions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

• Interactive and Gamified eBooks

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