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Emilia Ilieva, Reuben Matheka



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 $3 \text{ } 3 \text{Mg} + 1 \text{ } \text{Fe}_2\text{O}_3 \rightarrow 2 \text{ } \text{Fe} + 3 \text{ } \text{MgO}$ type of reaction single displacement
 $4 \text{ } 1 \text{C}_2\text{H}_4 + 3 \text{ } \text{O}_2 \rightarrow 2 \text{ } \text{CO}_2$

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 $\text{NaOH} + \text{KNO}_3 \rightarrow \text{NaNO}_3 + \text{KOH}$
 $\text{CH}_2\text{O} + \text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{Fe}_6\text{NaBr}_2 + \text{Fe}_2\text{Na}_3\text{Mg(OH)}_4 + 2 \text{ } 4 \text{ } \text{NH}_4\text{OH} + \text{HBr} + \text{H}_2\text{O} + \text{NH}_4\text{Br} + \text{PbO}_2 + \text{Na}_2\text{CO}_3 + \text{Na}_2\text{O} + \text{CO}_2 + \text{C}_4\text{H}_8\text{O}_6 + 4 \text{ } 2 \text{ } 2 \text{ } 4 \text{ } \text{H}_2\text{O} + \text{HCl} + \text{NaOH} + \text{H}_2\text{O} + \text{NaCl}$

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