Analytical Chemistry MCQs: TYBSC SEM III

1] Gravimetric Analysis

1. If the solubility of calcium fluoride in water is S moles per litre, its solubil is given by

(a) 2S
(b) $2S^2$
(b) $4S^3$
(d) $4S^2$

2. Correct representation of solubility product of Ca3(PO4)2 is

a) $[Ca^{2+}][PO^{3-}_4]^{2-}$
(b) $[Ca^{2+}][PO^{3-}]$
(c) [Ca] $[PO_4]$
(d) none of these

3. Dimethyl glyoxime is a reagent specific for the metal ion.

(a) Al
(b) Ni
(b) Fe
(d) Cu

4. The solubility product expression for Hg_2Cl_2 is

(a) $[Hg^{2+}]$ $[Cl^-]$
(b) $[Hg^{2+}_2][Cl^-]^{2-}$
c. $[Hg^{2+}_2][2Cl^-]$
(d) [Hg+1 [Cl]

5. A solution is called saturated if

(a) ionic product < solubility product.
(b) ionic product > solubility product
(c) ionic product = solubility product.
(d) none of these

6. Solubility of alcohol in water is due to

(a) ionic bonding
(b) covalent bonding
(c) Metallic bonding
(d) Hydrogen bonding

7. Precipitation takes place when the product of concentration of ions

(a) Exceeds their solubility product
(b) Is less than their solubility product
(c) Equal to their solubility product
(d) None of these

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Léo Pomar

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