[	Reactants	Products
(a)	H, Cl and HCl	2H, O, and H <sub>2</sub> O

Reactants	Products
HCl and 2H,O	H, Cl,2H and O,

	Reactants	Products
	H, Cl,2H and O <sub>2</sub>	HCl and 2H <sub>2</sub> O
)		

	Reactants	Products
	2H, O <sub>2</sub> and H <sub>2</sub> O	H, Cl and HCl
(db)		.,

Correct Answer: Option C

Q4) A student performs an experiment to form aluminium chloride from aluminium and chlorine. Which options give the chemical equation of the reaction?

- (a) Al + Cl<sub>2</sub> → AlCl<sub>2</sub>
- (b) 2Al + Cl<sub>i</sub> → 2AlCl
- (c) 2Al + 3Cl<sub>2</sub> → 2AlCl<sub>3</sub>
- (d) 3Al + 3Cl<sub>2</sub> → 3AlCl<sub>3</sub>

Correct Answer: Option C

Q5. Give the ratio in which hydrogen and oxygen are present in water by volume.

- (a) 1:2
- (b) 1:1
- (c) 2:1

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M Mark

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