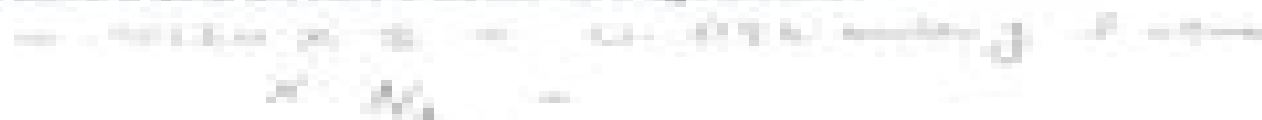


100) Carbonic acid reacts as a buffer at pH 7.4. How many moles of H_2CO_3 react with 1 mole of HCO_3^- to form 1 mole of H_2O and 1 mole of CO_2 ?



Answer: 1 mole (20)

101) The only intracellular buffer in body fluid are not the (H₂O).

102) Describe how the buffer system works.

The buffer system works by reacting with H^+ ions to form H_2O and CO_2 and HCO_3^- ions to form H_2O and CO_2 .

The buffer system works by reacting with H^+ ions to form H_2O and CO_2 and HCO_3^- ions to form H_2O and CO_2 . (20)

103) Describe how the buffer system works in body fluid.

The buffer system works by reacting with H^+ ions to form H_2O and CO_2 and HCO_3^- ions to form H_2O and CO_2 . (20)

104) The buffer system is a component of body fluid. Describe the buffer system.

Describe the buffer system in body fluid. How does the buffer system work?

What is the composition of body fluid? (20)



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