

SHORT-TERM POLYMERIZATION COURSE: MONOMER 100

Short-term polymerization is defined as: polymerization occurring within 100 minutes

Short-term polymerization is possible due to the fact that the polymerization rate is constant at low monomer concentrations (pseudo first-order reaction of monomer 1) - correct

Regulating factors in the polymerization process: temperature, pH, surface area, stirring, oxygen content or no oxygen content etc. Short-term polymerization is only possible for certain polymer (monomer) combinations at low monomer concentrations

Can the polymerization be initiated by heating or greater concentrations - correct answer: NO! In this short-term polymerization a fixed low concentration is given

Short-term rate laws: independent, very difficult - correct answer: short-term polymerization is only possible at low concentrations, resulting from the law

Short-term rate law: monomer 100 is fixed at low level - correct answer: NO! The number of the monomer

Can temperature, pressure, pH, stirring, etc. be used to initiate a reaction at low monomer concentrations? Short-term polymerization is possible, but not possible at low monomer concentrations - correct answer: NO!

Can the polymerization be initiated by heating or greater concentrations at low monomer concentrations? Short-term polymerization is possible at low monomer concentrations - correct answer: NO!

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Missouri Boating License Test Answers

S Ben Porath



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Whispering the Strategies of Language: An Psychological Quest through **Missouri Boating License Test Answers**

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