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S Baum



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web using this calculator you can calculate the molar volume of a gas for arbitrary temperature and pressure just note that for big values hundreds of atmospheres and thousands of degrees real gases divert from ideal gas law that's why they are not ideal and this formula can't be used

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web this volume is known as the molar gas volume V_m the volume of a gas may be calculated from its number of moles using volume of gas moles V_m rearranging this moles of gas $\frac{V}{V_m}$

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web jun 8 2023 $V = 1.00 \text{ mol} \times 0.08206 \text{ L atm mol}^{-1} \text{ K}^{-1} \times 273 \text{ K} = 22.4 \text{ L}$ in other words 1 mole of a gas will occupy 22.4 L at stp assuming ideal gas behavior at stp the volume of a gas is only dependent on number of moles of

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web aug 3 2022 volumes of gases and liquids are measured in cubic centimetres cm^3 or cubic decimetres dm^3 or litres l 1 mole of any gas at room temperature and pressure occupies 24 dm^3 $24\,000 \text{ cm}^3$ this value is known as the molar volume

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