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REPUBLIC OF SOUTH AFRICA

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PREPARATORY EXAMINATION
2016**

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**PHYSICAL SCIENCES: PHYSICS
FIRST PAPER**

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MARKS: 150

16 pages + 3 data sheets

PHYSICAL SCIENCES: Paper 1

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Table of Contents Gauteng Gr 10 Term4 Physical Science Paper

1. Understanding the eBook Gauteng Gr 10 Term4 Physical Science Paper
 - The Rise of Digital Reading Gauteng Gr 10 Term4 Physical Science Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Gauteng Gr 10 Term4 Physical Science Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gauteng Gr 10 Term4 Physical Science Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gauteng Gr 10 Term4 Physical Science Paper
 - Personalized Recommendations

- Gauteng Gr 10 Term4 Physical Science Paper User Reviews and Ratings
- Gauteng Gr 10 Term4 Physical Science Paper and Bestseller Lists
- 5. Accessing Gauteng Gr 10 Term4 Physical Science Paper Free and Paid eBooks
 - Gauteng Gr 10 Term4 Physical Science Paper Public Domain eBooks
 - Gauteng Gr 10 Term4 Physical Science Paper eBook Subscription Services
 - Gauteng Gr 10 Term4 Physical Science Paper Budget-Friendly Options
- 6. Navigating Gauteng Gr 10 Term4 Physical Science Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Gauteng Gr 10 Term4 Physical Science Paper Compatibility with Devices
 - Gauteng Gr 10 Term4 Physical Science Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gauteng Gr 10 Term4 Physical Science Paper
 - Highlighting and Note-Taking Gauteng Gr 10 Term4 Physical Science Paper
 - Interactive Elements Gauteng Gr 10 Term4 Physical Science Paper
- 8. Staying Engaged with Gauteng Gr 10 Term4 Physical Science Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gauteng Gr 10 Term4 Physical Science Paper
- 9. Balancing eBooks and Physical Books Gauteng Gr 10 Term4 Physical Science Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gauteng Gr 10 Term4 Physical Science Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gauteng Gr 10 Term4 Physical Science Paper
 - Setting Reading Goals Gauteng Gr 10 Term4 Physical Science Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gauteng Gr 10 Term4 Physical Science Paper
 - Fact-Checking eBook Content of Gauteng Gr 10 Term4 Physical Science Paper

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8.00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50.0 ml of water in the cylinder and add it to the beaker 4

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalise of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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