

Introduction to Organic Chemistry

Classifying and Naming Organic Compounds

Organic chemistry is the study of carbon and its compounds—their structures, properties and chemical reactions. A tremendous number of compounds come under this heading. There are more than nine million organic compounds and more than seven million organic reactions associated with them! All organic compounds contain carbon as well as hydrogen atoms attached to the carbon “skeleton” in predictable numbers. In most chemical reactions, the C—C skeleton does not change. Typically, organic reactions involve either C=C double or C≡C triple bonds in a molecule or carbon atoms attached to “heteroatoms,” such as oxygen, nitrogen or chlorine. Organic compounds are classified into functional group classes based on their structure and properties.

A *functional group* is defined as a specific arrangement of atoms, such as —OH or —NH₂, that is responsible for the types of reactions an organic compound will undergo. Functional groups are the reactive groups in a molecule. They undergo characteristic chemical reactions, such as oxidation or reduction, and also give related compounds similar physical properties. The study of organic functional groups provides the underlying basis for the science of organic chemistry. Knowing the chemistry of different functional groups allows chemists to explain and predict organic reactions and also to create new compounds.

Table 1 shows the structures of common organic functional groups. The symbol R is used to represent rings or chains of carbon atoms attached to the functional group.

Table 1. Structures of Organic Compounds and Functional Groups

Functional Group	Structure	Functional Group	Structure
Alkenes		Amines	$\text{R}-\text{NH}_2$ (or R_2NH and R_3N)
Alkynes		Ketones	
Aromatic Compounds		Aldehydes	
Alcohols	$\text{R}-\text{O}-\text{H}$	Carboxylic Acids	
Ethers	$\text{R}-\text{O}-\text{R}'$	Esters	
Alkyl Halides	$\text{R}-\text{X}$ where X = F, Cl, Br or I	Amides	

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