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Done
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    BACKGROUND-COLOR: transparent;
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Store Name	Order Date	Title	Qty	Total
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Barnum's	9/13/1994	Is Anger the Enemy?	75	821.25
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Bookbeat	9/14/1994	The Busy Executive's Database	10	199.9
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Bookbeat	5/22/1993	But Is It User Friendly?	30	688.5
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Microsoft Access Report Xml Stylesheet

Julitta Korol



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Microsoft Office Access 2007 Inside Out Jeff Conrad,John L. Viescas,2007-04-11 You re beyond the basics so dive right in and really put

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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton $HA + B \rightleftharpoons H^+ + B^-$ strong acids e.g. sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral ph at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web 1 approximately 3 g of hydrated oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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