

34. The maximum number of covalent bonds formed by nitrogen is

A. 1.  
B. 2.  
C. 3.  
D. 4.

35. The IUPAC name of the compound  $\text{CH}_3\text{CH}(\text{CH}_3)\text{CHCH}_2$  is

A. 2 - methyl but - 1 - ene.  
B. 2 - methyl but - 2 - ene.  
C. 3 - methyl but - 1 - ene.  
D. 3 - methyl but - 2 - ene.

36. Ionization energy increases across the period in the periodic table because

A. atomic number increases.  
B. effective nuclear charge increases.  
C. mass number decreases.  
D. screening effect decreases.

37. Which of the following properties indicate that an element is a metal. It

I. reacts with oxygen to form an acidic oxide  
II. forms ionic chlorides.  
III. has variable oxidation states.  
IV. displaces hydrogen from dilute HCl.

A. I and III only  
B. I and II only  
C. II and IV only  
D. I, II, III and IV

38. The electron configuration of carbon atom in its excited state is

A.  $1s^2 2s^2 2p_x^2 2p_y^0$   
B.  $1s^2 2s^2 2p_x^1 2p_y^1$   
C.  $1s^2 2s^1 2p_x^1 2p_y^1 2p_z^1$   
D.  $1s^1 2s^2 2p_x^1 2p_y^1 2p_z^1$

39. An oxide has the following properties. It

I. is a white powder.  
II. reacts with HCl.  
III. reacts with NaOH.  
IV. is insoluble in water.

The oxide is

A. alkaline.  
B. amphoteric.  
C. acidic.  
D. neutral.

40. Which of the following statements about atoms of a metal is correct? They

A. readily accept electrons.  
B. are soft.  
C. are held together by covalent bond.  
D. are held together by a sea of electron cloud.

41. Which of the following sketches is a graphical representation of Boyle's law?



42. The atom with the electron configuration  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^4$  is in

A. period 4, p- block.  
B. period 3, p- block.  
C. period 4, d- block.  
D. period 3, d- block.

# May 2013 Chemistry C

**Chaudhery Mustansar Hussain, Boris  
Kharisov**



## May 2013 Chemistry C:

Chemistry Education Javier García-Martínez, Elena Serrano-Torregrosa, 2015-02-17 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry students Electron Paramagnetic Resonance Bruce Gilbert, Damien M Murphy, Victor Chechik, 2014-09-30 The topics covered in this series describe contrasting types of Electron Paramagnetic Resonance EPR application with results being set into the context of earlier work and presented as a set of critical yet coherent overviews *Photochemistry* Elisa Fasani, Angelo Albini, 2014-08-19 Reviewing photo induced processes that have relevance to a wide ranging number of academic and commercial disciplines and interests covering chemistry physics biology and technology this series is essential reading for anyone wishing to keep abreast of the current literature Now in its 42nd volume and with contributions from across the globe this series continues to present an accessible digest of current opinion and research in all aspects of photochemistry Topics covered in this volume include the state of the art in computational photochemistry advances in dye sensitized photopolymerization processes photoclick chemistry and continuous flow photochemical reactions This Specialist Periodical Report presents critical and comprehensive reviews of the last 12 months of the primary literature drawing on 100 s of citations and is an essential resource for anyone working at the cutting edge of photochemistry and a gateway to newcomers in the field **Heterogeneous Gold Catalysts and Catalysis** Zhen Ma, Sheng Dai, 2014-09-10 Once considered an inert element gold has recently gained attention as a catalyst With hundreds of papers being published each year this book presents a comprehensive review of this rapidly evolving field with contributions by leading experts across the globe Going through the chapters citing the primary literature the reader will gain a thorough background to the use of gold in catalysis as well as the latest methods for the preparation of gold catalysts Other chapters demonstrate the characterisation and modelling of gold catalysed reactions with consideration given to both the fundamentals and commercial applications of this emerging group of catalysts Written to be accessible by postgraduates and newcomers to the field this book will also benefit experienced researchers and therefore be an essential reference in the

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