- What are molecules? Give examples.
- What is a physical change and littlesamples of physical changes.
- What is a chemical change and give examples of chemical changes.

2.2 Main Rody (Leucon per sentializa)

Educator starts lesson off with an exciting demonstration as mentioned in the introduction and explains the relevant concepts in terms of a chemical change. [20 min.]

- Matter it all around us and it undergoes changes all the time. Mese changes can be classified as PRYSICAL CHANGES OR CHEMICAL CHANGES.
- The CONCEPTS of PHYSICAL CHANGE and CHEMICAL CHANGE is shown in the table below, educator engages learners in a discussion of the different aspects of physical and chemical change as mentioned in the table to re-enforce the concepts.

physical changes	e beminal e hanges
common signs that a chemical change has occurred are:	common signs that a physical change has occurred are:
I. Production of gas bubbles C. Change in the way something smelt I. A release of energy such as a flash or a sound (like a firecracker) I. A precipitate forms (two liquids mixed together form a solid and a liquid)	I. change in the size I. change in shape. I. change in colour, or I. Change in state/ phase of matter of a substance. I. No new substance is produced.
humples of physical changes	Facompile a of e he mic al e hange a
A grape when Hepped on (changes shape) Blowing up a balloon (changes size and shape) Liquid water furning to ice (changes state of matter) Liquid water furning to steam (changes state of matter) Missing salt and sugar (changes the appearance, but you can still reparate the mixture) Missing water and salt (changes the appearance, but you can still reparate the mixture)	Metal sisting prew substance formest; Stamach algesting food (break down of food to new substances) Plant's anying out photosynthesis (putting water and carbon dioxide together to make sugar) Mixing boiling sodo and vinegar (makes a neutral liquid and a gas)

- Educator documes the results of the experiment with the learners.
- The liquid hydrogen peroxide decomposes (breaks up) to hirm oxygen gas and liquid water (the atomic model the using ball and stock as well as space filling diagrams to show the reactants and products in this reaction).
- The MANGANESE DIGKDE is a catalyst and speeds up the reaction. A CARDING a chemical substance that changes the speed of a reaction without getting used up in the
 process, of the end of the reaction the catalyst can be recovered. Eg. ENCYMES in our body help to speed up the digestive process.

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