

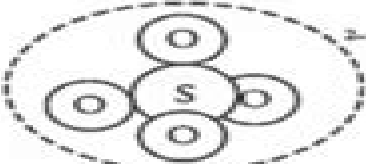
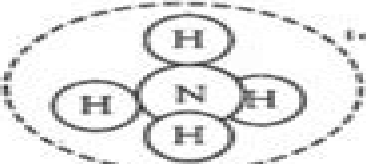
Polyatomic Ions

Can a group of atoms have a charge?

Why?

Do you know you eat a lot of "-ates"? Next time you look at a food label, read the ingredients and you will likely find a number of ingredients that end with "-ate," such as sodium phosphate or calcium carbonate. Did you ever wonder what the chemical formulas of these ingredients look like? In this activity we will explore polyatomic ions, which are groups of atoms that carry a charge. These ions are found in our food ingredients, natural waterways, and many other chemical compounds you encounter every day.

Model 1 – Types of Ions

Monatomic Ions	Nitride 	Sulfide 	Chloride 
Polyatomic Ions	Nitrate 	Sulfate 	Ammonium 
	Nitrite 	Sulfite 	Hydroxide 

1. Use Model 1 to complete the table below.

Name of Ion	Nitride	Nitrate	Sulfate	Sulfite	Ammonium
Charge on Ion	-3	-1	-2	-2	+1
Type and Number of Atoms	1 nitrogen	1 nitrogen 3 oxygen	1 sulfur 4 oxygen	1 sulfur 3 oxygen	1 nitrogen 4 hydrogen
Chemical Formula	N^{3-}	NO_3^{1-}	SO_4^{2-}	SO_3^{2-}	NH_4^{1+}

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Table of Contents Flinn Scientific Pogil Activity Polyatpmic Ions

1. Understanding the eBook Flinn Scientific Pogil Activity Polyatpmic Ions
 - The Rise of Digital Reading Flinn Scientific Pogil Activity Polyatpmic Ions
 - Advantages of eBooks Over Traditional Books
2. Identifying Flinn Scientific Pogil Activity Polyatpmic Ions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Flinn Scientific Pogil Activity Polyatpmic Ions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Flinn Scientific Pogil Activity Polyatpmic Ions
 - Personalized Recommendations
 - Flinn Scientific Pogil Activity Polyatpmic Ions User Reviews and Ratings
 - Flinn Scientific Pogil Activity Polyatpmic Ions and Bestseller Lists

5. Accessing Flinn Scientific Pogil Activity Polyatpmic Ions Free and Paid eBooks
 - Flinn Scientific Pogil Activity Polyatpmic Ions Public Domain eBooks
 - Flinn Scientific Pogil Activity Polyatpmic Ions eBook Subscription Services
 - Flinn Scientific Pogil Activity Polyatpmic Ions Budget-Friendly Options
6. Navigating Flinn Scientific Pogil Activity Polyatpmic Ions eBook Formats
 - ePub, PDF, MOBI, and More
 - Flinn Scientific Pogil Activity Polyatpmic Ions Compatibility with Devices
 - Flinn Scientific Pogil Activity Polyatpmic Ions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Flinn Scientific Pogil Activity Polyatpmic Ions
 - Highlighting and Note-Taking Flinn Scientific Pogil Activity Polyatpmic Ions
 - Interactive Elements Flinn Scientific Pogil Activity Polyatpmic Ions
8. Staying Engaged with Flinn Scientific Pogil Activity Polyatpmic Ions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Flinn Scientific Pogil Activity Polyatpmic Ions
9. Balancing eBooks and Physical Books Flinn Scientific Pogil Activity Polyatpmic Ions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Flinn Scientific Pogil Activity Polyatpmic Ions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Flinn Scientific Pogil Activity Polyatpmic Ions
 - Setting Reading Goals Flinn Scientific Pogil Activity Polyatpmic Ions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Flinn Scientific Pogil Activity Polyatpmic Ions
 - Fact-Checking eBook Content of Flinn Scientific Pogil Activity Polyatpmic Ions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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