STOICHIOMETRY: MOLE-MOLE PROBLEMS

Name ____

1. N₂ + 3H₂ → 2NH₃

How many moles of hydrogen are needed to completely react with two moles of nitrogen?

2. 2KCIO, → 2KCI + 3O₂

How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate?

3. Zn + 2HCl → ZnCl, + H,

How many moles of hydrogen are produced from the reaction of three moles of zinc with an excess of hydrochloric acid?

4. C₃H_a + 5O₂ → 3CO₃ + 4H₂O

How many moles of oxygen are necessary to react completely with four moles of propane (C₃H₈)?

K₃PO₄ + Al(NO₅)₅ → 3KNO₅ + AlPO₄

How many moles of potassium nitrate are produced when two moles of potassium phosphate react with two moles of aluminum nitrate?

Episode 1the Mole Answers

WJ Hussar



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