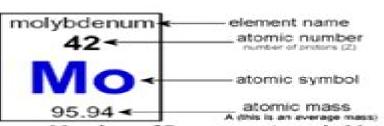
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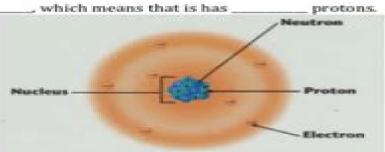
SPS1: Students will investigate our current understanding of the atom.

a. Examine the structure of the atom in terms of

- proton, electron, and neutron locations.
 - Proton has a charge of ______ and is located in the ______ of the atom.
 - Protons are also the elements number of the identity of the element, so protons are found by looking at the atomic number
 - Neutron has a charge of ______ and is located in the _____ of the atom.
 - Neutrons are found by subtracting the ______ Number from the ______ Mass or the BIG #-SMALL #.
 - Electron has a charge of _____ and is located in the ____ of the atom.
 - Electrons are equal to the number of ______ in a neutral atom.
 - The nucleus is made up of these 2 particles______ and _____
 - The electrons are located in this region of the atom, known as the ______
 - The electron cloud is divided in to energy levels or energy shells, the first level will hold electrons and the second level will hold ______ electrons.
 - The valence shell or valence level is the ______ energy shell of the atom.
- atomic mass and atomic number.
 - atoms identity-it's the small number of the periodic table.
 - Atomic mass is equal to the average mass of all the isotopes of a particular element—it the big number on the periodic table.
- atoms with different numbers of neutrons (isotopes).
 - Define Isotope
 - Ex. Carbon-12 and Carbon-14
 - Carbon 12 has ____P ___N ___E Carbon 14 has ___P ___N ___E
- explain the relationship of the proton number to the element's identity.
 - The ______ number is equal to the number of _____ in the atom which is how the element is identified.
 - Ex. Nitrogen has an atom number of , which means that is has protons.



- Number of Protons = Atomic Number
- Number of Electrons = Number of Protons = Atomic Number
- Number of Neutrons = Mass Number Atomic Number



Eoct Physics Study Guide

Rosina Ehmann

Eoct Physics Study Guide:

Eoct Physics Study Guide Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Eoct Physics Study Guide**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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