

Binary Ionic Compounds

Name: _____ Per: _____

Directions: First quickly scan the worksheet and circle any metals (as a symbol or as a name) that is a transition metal. Then either give the same or chemical formula as necessary for each problem below.

Remember that transition metals can have multiple oxidation states, so you are required to indicate the appropriate oxidation state in parenthesis when naming them. You must use the subscript and oxidation state of the anion to help determine the oxidation state of the transition metal.

- | | |
|--|-----------------------------------|
| 1. LiF _____ | 2. lithium chloride _____ |
| 3. Li ₂ O _____ | 4. lithium nitride _____ |
| 5. Li ₃ P _____ | 6. beryllium fluoride _____ |
| 7. BeO _____ | 8. beryllium sulfide _____ |
| 9. BF ₃ _____ | 10. boron chloride _____ |
| 11. BBr ₃ _____ | 12. boron oxide _____ |
| 13. BN _____ | 14. sodium fluoride _____ |
| 15. CuF _____ | 16. copper (II) chloride _____ |
| 17. Cu ₂ O _____ | 18. copper (II) oxide _____ |
| 19. Cu ₃ N _____ | 20. copper (II) nitride _____ |
| 21. Na ₂ O _____ | 22. sodium nitride _____ |
| 23. PbF ₂ _____ | 24. lead (IV) fluoride _____ |
| 25. PbS _____ | 26. lead (IV) sulfide _____ |
| 27. Pb ₃ N ₄ _____ | 28. lead (IV) oxide _____ |
| 29. FeF ₃ _____ | 30. iron (II) bromide _____ |
| 31. Fe ₂ O ₃ _____ | 32. chromium (VI) phosphide _____ |
| 33. FeP _____ | 34. aluminum fluoride _____ |
| 35. AlI ₃ _____ | 36. aluminum oxide _____ |
| 37. Mn ₂ O ₇ _____ | 38. manganese (IV) bromide _____ |
| 39. SnSe _____ | 40. tin (IV) oxide _____ |

Ionic Compound Worksheets With Answers

Arshad Iqbal



Ionic Compound Worksheets With Answers:

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properties of group VII elements reaction of group II elements with oxygen reactions of group II elements reactions of group VII elements thermal decomposition of carbonates and nitrates thermal decomposition of group II carbonates thermal decomposition of group II nitrates uses of group II elements uses of group II metals uses of halogens and their compounds The Halogenoalkanes MCQ PDF e Book Chapter 14 practice test to solve MCQ questions on Halogenoalkanes uses of halogenoalkanes elimination reactions nucleophilic substitution in halogenoalkanes and nucleophilic substitution reactions The Hydrocarbons MCQ PDF e Book Chapter 15 practice test to solve MCQ questions on Introduction to alkanes sources of alkanes addition reactions of alkenes alkane reaction alkenes and formulas The Introduction to Organic Chemistry MCQ PDF e Book Chapter 16 practice test to solve MCQ questions on Organic chemistry functional groups organic reactions naming organic compounds stereoisomerism structural isomerism and types of organic reactions The Ionic Equilibria MCQ PDF e Book Chapter 17 practice test to solve MCQ questions on Introduction to ionic equilibria buffer solutions equilibrium and solubility indicators and acid base titrations pH calculations and weak acids The Lattice Energy MCQ PDF e Book Chapter 18 practice test to solve MCQ questions on Introduction to lattice energy ion polarization lattice energy value atomization and electron affinity Born Haber cycle and enthalpy changes in solution The Moles and Equations MCQ PDF e Book Chapter 19 practice test to solve MCQ questions on Amount of substance atoms molecules mass chemical formula and equations gas volumes mole calculations relative atomic mass solutions and concentrations The Nitrogen and Sulfur MCQ PDF e Book Chapter 20 practice test to solve MCQ questions on Nitrogen gas nitrogen and its compounds nitrogen and gas properties ammonia ammonium compounds environmental problems caused by nitrogen compounds and nitrate fertilizers sulfur and oxides sulfuric acid and properties and uses of sulfuric acid The Organic and Nitrogen Compounds MCQ PDF e Book Chapter 21 practice test to solve MCQ questions on Amides in chemistry amines amino acids peptides and proteins The Periodicity MCQ PDF e Book Chapter 22 practice test to solve MCQ questions on Acidic oxides basic oxides aluminum oxide balancing equation period 3 chlorides balancing equations reactions with chlorine balancing equations reactions with oxygen bonding nature of period 3 oxides chemical properties of chlorine chemical properties of oxygen chemical properties periodicity chemistry periodic table chemistry oxides chlorides of period 3 elements electrical conductivity in period 3 oxides electronegativity of period 3 oxides ionic bonds molecular structures of period 3 oxides oxidation number of oxides oxidation numbers oxides and hydroxides of period 3 elements oxides of period 3 elements period III chlorides periodic table electronegativity physical properties periodicity reaction of sodium and magnesium with water and relative melting point of period 3 oxides The Polymerization MCQ PDF e Book Chapter 23 practice test to solve MCQ questions on Types of polymerization polyamides polyesters and polymer deductions The Rates of Reaction MCQ PDF e Book Chapter 24 practice test to solve MCQ questions on Catalysis collision theory effect of concentration reaction kinetics and temperature effect on reaction rate The Reaction Kinetics MCQ PDF e Book Chapter 25 practice test to solve MCQ questions on Reaction kinetics

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Table of Contents Ionic Compound Worksheets With Answers

1. Understanding the eBook Ionic Compound Worksheets With Answers
 - The Rise of Digital Reading Ionic Compound Worksheets With Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Compound Worksheets With Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Compound Worksheets With Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Compound Worksheets With Answers
 - Personalized Recommendations
 - Ionic Compound Worksheets With Answers User Reviews and Ratings
 - Ionic Compound Worksheets With Answers and Bestseller Lists

5. Accessing Ionic Compound Worksheets With Answers Free and Paid eBooks
 - Ionic Compound Worksheets With Answers Public Domain eBooks
 - Ionic Compound Worksheets With Answers eBook Subscription Services
 - Ionic Compound Worksheets With Answers Budget-Friendly Options
6. Navigating Ionic Compound Worksheets With Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Compound Worksheets With Answers Compatibility with Devices
 - Ionic Compound Worksheets With Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Compound Worksheets With Answers
 - Highlighting and Note-Taking Ionic Compound Worksheets With Answers
 - Interactive Elements Ionic Compound Worksheets With Answers
8. Staying Engaged with Ionic Compound Worksheets With Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Compound Worksheets With Answers
9. Balancing eBooks and Physical Books Ionic Compound Worksheets With Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Compound Worksheets With Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ionic Compound Worksheets With Answers
 - Setting Reading Goals Ionic Compound Worksheets With Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ionic Compound Worksheets With Answers
 - Fact-Checking eBook Content of Ionic Compound Worksheets With Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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