

<b>Activity A:</b> <b>Small atoms</b>	<b>Get the Gizmo ready:</b> <ul style="list-style-type: none"> <li>On the PERIODIC TABLE tab, select H (hydrogen).</li> <li>Select the ELECTRON CONFIGURATION tab.</li> <li>Click <b>Reset</b>.</li> </ul>	
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**Introduction:** Electrons are arranged in **orbitals**, **subshells**, and **shells**. These levels of organization are shown by the boxes of the Gizmo. Each box represents an orbital. The subshells are labeled with letters (*s*, *p*, *d*, and *f*) and the shells are labeled with numbers.

**Question:** How are electrons arranged in elements with atomic numbers 1 through 10?

1. Infer: Based on its atomic number, how many electrons does a hydrogen atom have? \_\_\_\_\_

2. Arrange: The **Aufbau principle** states that electrons occupy the lowest-energy orbital. Click once in the **1s** box to add an electron to the only orbital in the *s* subshell of the first shell.

Click **Check**. What is the electron configuration of hydrogen? \_\_\_\_\_

3. Arrange: Click **Next element** to select helium. Add another electron to the **1s** orbital. The arrows represent the **spin** of the electron. What do you notice about the arrows?

\_\_\_\_\_

The **Pauli exclusion principle** states that electrons sharing an orbital have opposite spins.

4. Check your work: Click **Check**. What is the electron configuration of helium? \_\_\_\_\_

5. Arrange: Click **Next element** and create electron configurations for lithium, beryllium, and boron. Click **Check** to check your work, and then list each configuration below:

Lithium: \_\_\_\_\_ Beryllium: \_\_\_\_\_ Boron: \_\_\_\_\_

6. Arrange: Click **Next element** to select carbon. Add a second electron to the first **2p** orbital.

Click **Check**. What feedback is given? \_\_\_\_\_

7. Rearrange: **Hund's rule** states that electrons will occupy an empty orbital when it is available in that subshell. Rearrange the electrons within the **2p** subshell and click **Check**.

Is the configuration correct now? \_\_\_\_\_

Show the correct configuration in the boxes at right:

1s ☐

2s ☐ 2p ☐ ☐ ☐

(Activity A continued on next page)

# Ionic Bonds Explore Learning Answer Key

**Kathleen Armour**



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## **Table of Contents Ionic Bonds Explore Learning Answer Key**

1. Understanding the eBook Ionic Bonds Explore Learning Answer Key
  - The Rise of Digital Reading Ionic Bonds Explore Learning Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Bonds Explore Learning Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ionic Bonds Explore Learning Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Bonds Explore Learning Answer Key
  - Personalized Recommendations
  - Ionic Bonds Explore Learning Answer Key User Reviews and Ratings
  - Ionic Bonds Explore Learning Answer Key and Bestseller Lists
5. Accessing Ionic Bonds Explore Learning Answer Key Free and Paid eBooks
  - Ionic Bonds Explore Learning Answer Key Public Domain eBooks
  - Ionic Bonds Explore Learning Answer Key eBook Subscription Services
  - Ionic Bonds Explore Learning Answer Key Budget-Friendly Options
6. Navigating Ionic Bonds Explore Learning Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Ionic Bonds Explore Learning Answer Key Compatibility with Devices
- Ionic Bonds Explore Learning Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Ionic Bonds Explore Learning Answer Key
  - Highlighting and Note-Taking Ionic Bonds Explore Learning Answer Key
  - Interactive Elements Ionic Bonds Explore Learning Answer Key
- 8. Staying Engaged with Ionic Bonds Explore Learning Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ionic Bonds Explore Learning Answer Key
- 9. Balancing eBooks and Physical Books Ionic Bonds Explore Learning Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Ionic Bonds Explore Learning Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Ionic Bonds Explore Learning Answer Key
  - Setting Reading Goals Ionic Bonds Explore Learning Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ionic Bonds Explore Learning Answer Key
  - Fact-Checking eBook Content of Ionic Bonds Explore Learning Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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