

## CHAPTER 7

## STUDY GUIDE

## Ionic Compounds and Metals

### Section 7.1 Ion Formation

*In your textbook, read about chemical bonds and formation of ions.*

Use each of the terms below just once to complete the passage.

chemical bond	electrons	energy level	ions	noble gases
nucleus	octet	pseudo-noble gas formations	valence	

The force that holds two atoms together is called a(n) (1) \_\_\_\_\_.  
Such an attachment may form by the attraction of the positively charged  
(2) \_\_\_\_\_ of one atom for the negatively charged  
(3) \_\_\_\_\_ of another atom, or by the attraction of charged atoms, which are called  
(4) \_\_\_\_\_. The attractions may also involve  
(5) \_\_\_\_\_ electrons, which are the electrons in the outermost  
(6) \_\_\_\_\_. The (7) \_\_\_\_\_ are a family of elements that have very  
little tendency to react. Most of these elements have a set of eight outermost electrons, which is called a  
stable (8) \_\_\_\_\_. The relatively stable electron structures developed by loss of  
electrons in certain elements of groups 3, 4, 13, and 14 are called (9) \_\_\_\_\_.

For each statement below, write *true* or *false*.

- \_\_\_\_\_ 10. A positively charged ion is called an anion.
- \_\_\_\_\_ 11. Elements in group 1 lose their one valence electron, forming an ion with a 1+ charge.
- \_\_\_\_\_ 12. Elements tend to react so that they acquire the electron structure of a halogen.
- \_\_\_\_\_ 13. A sodium atom tends to lose one electron when it reacts.
- \_\_\_\_\_ 14. The electron structure of a zinc ion ( $\text{Zn}^{2+}$ ) is an example of a pseudo-noble gas formation.
- \_\_\_\_\_ 15. A  $\text{Cl}^-$  ion is an example of a cation.
- \_\_\_\_\_ 16. The ending *-ide* is used to designate an anion.
- \_\_\_\_\_ 17. Nonmetals form a stable outer electron configuration by losing electrons and becoming anions.

# Ionic Compounds And Metals Study Guide Answers

**Hongru Du**



## **Ionic Compounds And Metals Study Guide Answers:**

## Reviewing **Ionic Compounds And Metals Study Guide Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Ionic Compounds And Metals Study Guide Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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