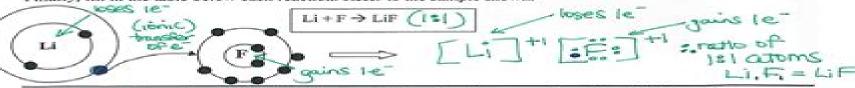
# Ponic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms Valence electrons Electron transfer from/to each atom Ions formed in the product Li Ive Houster's Ive to F. Lit Free accepts Ive from Li F-

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	loss formed in the product
Li +ci→ Lici  Li +ci→ Lici  Li +ci→ Lici	Li	Ive	donate le-	Lit
	CI	lve	accepts	C1-1
2) Ca+0 => Ca0  (a) + 3 = 0 => [a] +2 [0] - 2	Ca	are	Stereto -26	Cata
	0	6ve	accept	0-
3) Be + F => BeF <sub>2</sub> Be + F: -> [Be] + F: ]	Re	ave	donates ale to each F	Beto
	F	702	each F accepts Ive each	
mg + 5 -> Mgs mg + 5 -> [mg]+2[:S:] <sup>2</sup>	mg	ave	denotes ae to	wats
	S	Gve	acepts	5-3
S)K+F→KF  KU+FF: → [K]+[F]	K	Ive	donates	K+I
	F	7ve	accepts	F-/

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