

Chemical Bonding Review

Chemical Bonding Review

Name: _____

Directions: Answer each of the following questions by circling the correct answer choice.

- Which two elements have the same number of valence electrons?
A. Boron and Aluminum
B. Carbon and Nitrogen
C. Argon and Helium
D. Silicon and Aluminum
- What type of bond forms when electrons are transferred?
A. Covalent
B. Ionic
C. Metallic
D. Nonmetallic
- What is the chemical symbol for a potassium ion?
A. K^{+1}
B. K^{-1}
C. P^{+1}
D. P^{-1}
- How many electrons are needed for a filled outermost energy level in the Lewis-dot structure below?



- A. 4
B. 0
C. 2
D. 1
- When atoms gain electrons during chemical bonding, they become
A. Positively charged ions
B. Neutral ions
C. Negatively charged ions
D. Chemically charged ions
 - Potassium chloride is an example of what type of bond?
A. Metallic bond
B. Colloid bond
C. Covalent bond
D. Ionic bond
 - The diagram below shows what type of bond?



- A. Ionic
B. Covalent
C. Metallic
D. Solution

Ionic And Metallic Bonding Test B Answers

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Ionic And Metallic Bonding Test B Answers:

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