

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Number of Electrons Lost or Gained	Charge
1	Sodium	Na^+	11	10	lost one	positive
2	Chlorine	Cl^-	17	18	gained one	negative
3	Calcium	Ca^{2+}	20	18	lost two	positive
4	Oxygen	O^{2-}	8	10	gained two	negative
5	Aluminum	Al^{3+}	13	10	lost three	positive
6	Sulfur	S^{2-}	16	18	gained two	negative
7	Hydrogen	H^+	1	0	lost one	positive
8	Lithium	Li^+	3	2	lost one	positive
9	Fluorine	F^-	9	10	gained one	negative
10	Magnesium	Mg^{2+}	12	10	lost two	positive
11	Strontium	Sr^{2+}	38	36	lost two	positive
12	Barium	Ba^{2+}	56	54	lost two	positive
13	Strontium	Sr^{2+}	38	36	lost two	positive
14	Calcium	Ca^{2+}	20	18	lost two	positive
15	Sodium	Na^+	11	10	lost one	positive
16	Chlorine	Cl^-	17	18	gained one	negative
17	Fluorine	F^-	9	10	gained one	negative
18	Lithium	Li^+	3	2	lost one	positive
19	Sodium	Na^+	11	10	lost one	positive
20	Calcium	Ca^{2+}	20	18	lost two	positive

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RS Peters



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