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web log  $10^{-4}$   $0.5 \times 10^{-5}$  mol dm<sup>-3</sup> log  $\text{pK}_a$   $\text{pH}$  3.75 4.05 0.30 10.0 30.0 50.0 and  $\text{HCOOH}$  0.1 so  $\text{HCOO}^-$  0.200 mol dm<sup>-3</sup> amount vol x

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web nov 5 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low  $\text{pH}$

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web section 2  $\text{pH}$  of strong acids number of protons released monoprotic acid acid that releases one  $\text{H}^+$  ion per molecule e g  $\text{HCl}$  hydrochloric acid  $\text{HNO}_3$  nitric acid ch

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web jan 10 2021  $\text{S}^{2-}$   $\text{S}^{2-}$  strong electrolytes are 100 ionized and as long as the component ions are neither weak acids nor weak bases the ionic species present result

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web problem 1 which of the following are not acids  $\text{CH}_3\text{COOH}$   $\text{CO}_2$   $\text{HNO}_2$   $\text{HCOOH}$   $\text{CCl}_4$   $\text{CH}_3\text{COOH}$   $\text{CO}_2$   $\text{HNO}_2$   $\text{HCOOH}$   $\text{CCl}_4$  shazia naz numerade educator 02 23

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