# Analytical Chemistry MCQ Test. September'21.

- The metal hydroxide which is insoluble in caustic soda but soluble in excess of ammonium hydroxide.
- ammonium hydroxide.

  1. Cu(OH): 2 Zn(OH): 3 Fe(OH): 4 None
- 2. The salt solution which does not react with ammonium hydroxide is ------
- 1. Calcium nitrate 2. Zinc nitrate
- 3.Lead nitrate 4. Copper nitrate
- 3. The metal oxide which can react with acids as well as alkali is------
  - 1. Silver oxide 2. Copper (II) oxide
  - 3. Aluminum oxide 4. Calcium oxide
  - 4.Al metal shows amphoteric behaviour.it can be shown by
  - 1. Its reaction with dil.NaOH only.
  - Its reaction with dil.HCl only.
  - 3. Its reaction with both dil.HCl and dil.NaOH.
  - It does not react with either dil.HCl or dil.NaOH

#### QUESTION 5

Which of the following hydroxides soluble in excess of NaOH?

- 1. Cu (OH)<sub>2</sub> 2. Zn (OH)<sub>2</sub> 3. Fe (OH)<sub>2</sub> 4Fe(OH)<sub>3</sub>
- The salt which gives a dirty green precipitate with sodium hydroxide solution is.
  - 1. Iron (III) sulphate 2. Iron (II) sulphate
  - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a reddish-brown precipitate with sodium hydroxide solution is.
  - 1. Iron (III) sulphate 2. Iron (II) sulphate
  - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a gelatinous white precipitate with Ammonium hydroxide solution is.
  - 1. Iron (III) choride 2. Iron (II) sulphate
  - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a chalky white precipitate with Ammonium hydroxide solution is.
  - 1. Lead nitrate 2. Iron (II) sulphate
  - 3.Copper sulphate 4. Zinc sulphate

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**Basic Concepts of Chemistry and MCQ for NEET** 

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