

Acidity of Beverages

AP® Chemistry Big Idea 1: Investigation 4 An Advanced Inquiry Unit

Introduction

Common beverages (soft drinks, sports drinks, fruit juices, teas, coffees) are acidic because their water-based solutions are rich with dissolved substances including acids. In this investigation, students will use pH paper and indicators to determine the concentrations of acids in beverages (primarily soft, fruit-flavored, sports drinks).

Concepts

- Acid-base levels
- Indicators

- Buffers
- pH

- Equilibrium
- Concentration

Background

Most acids will ionize in water, and these acids are often used as strong acids in water and the resulting solution is acidic. Acids that have low dissociation constants are



Biphenyl, organic compounds and their salts

The structure of biphenyl is shown below for background information. It is a representative of aromatic hydrocarbons, or benzene derivatives. In this investigation, students will use pH paper and indicators to determine the concentrations of acids in beverages (primarily soft, fruit-flavored, sports drinks).



Acid-base chemistry is an important concept in many areas of chemistry and is often used in a variety of applications. The following table is provided for background information and is not intended to be used as a reference.

In the laboratory, students will use pH paper and indicators to determine the concentrations of acids in beverages (primarily soft, fruit-flavored, sports drinks). Students will use pH paper and indicators to determine the concentrations of acids in beverages (primarily soft, fruit-flavored, sports drinks).

Flinn Scientific Inc Acidity Of Beverages Answers

M Carnoy



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