	レビン	
Name:	KU /	Block:

# Assignment #2 – Compound Names and Formulas Multi-valent ions only

A. Write the correct formula for the	e following compounds, all of which have been named using the
	The ionic charge is given after the first element (metallic) in Roman
Numerals.	

1. copper (II) oxide CUO	11. manganese (III) oxide Mv 203		
2. mercury (I) oxide HQ20	12. vanadium (II) bromide VB(z		
3. gold (III) chloride _AuCl3	13. niobium(V) oxide Nb205		
4. thallium (III) bromide TIPY 3	14. titanium (III) oxide		
5. bismuth (V) oxide	15. titanium (III) nitride TIN		
6. terbium (III) oxide	16. iron (II) oxide <u>Fe.O</u>		
7. uranium (VI) oxide \( \lambda \ \O \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	17. cobalt (II) phosphide C03P2		
8. protactinium (IV) bromide PhBY4	18. tin (II) oxide		
9. cerium (III) oxide	19. thulium (II) bromide TWBY 2		
10. arsenic (V) sulphideAS_2S_5	20. copper (II) bromide CUPY2		
B. Name the compound using the Roman Numeral Method and balance the ionic charges.			
1. SnCl Tin (IV) Chloride	9. Pdf. Palladium (IV) Fluoride		
2. BIBIS BISMINTO STOMICE	10. OszOs OSMIUM (III) Oxide		
3. POO2 ROLONIUM (IV) OXICE	11. MOBEZ MOLYBORENIUM (II) Bramide		
4. Pbl2 100d(II) Iodide	12. VCIs Vanadium (V) Chloride		
5. HgO <u>Mercury</u> (II) Oxide	13. Mn2O3 Manganese (III) Oxide		
6. HgCI MEYCUYY (I) Chroride	14. COO COLONY (II) Oxide		
7. Auzos Gold (III) Oxide	15. Np2O3 NEDTUNIUM (II) OXICE		
8. FeClo Iron (II) Chloride	16. V2O6 Vanadium (V) Oxide		

# **Naming Balancing And Formula Writing Answer Key**

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#### **Table of Contents Naming Balancing And Formula Writing Answer Key**

- 1. Understanding the eBook Naming Balancing And Formula Writing Answer Key
  - The Rise of Digital Reading Naming Balancing And Formula Writing Answer Key
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Naming Balancing And Formula Writing Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Naming Balancing And Formula Writing Answer Key
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Naming Balancing And Formula Writing Answer Key
  - Personalized Recommendations
  - Naming Balancing And Formula Writing Answer Key User Reviews and Ratings
  - Naming Balancing And Formula Writing Answer Key and Bestseller Lists
- 5. Accessing Naming Balancing And Formula Writing Answer Key Free and Paid eBooks
  - Naming Balancing And Formula Writing Answer Key Public Domain eBooks
  - Naming Balancing And Formula Writing Answer Key eBook Subscription Services
  - Naming Balancing And Formula Writing Answer Key Budget-Friendly Options
- 6. Navigating Naming Balancing And Formula Writing Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Naming Balancing And Formula Writing Answer Key Compatibility with Devices
  - Naming Balancing And Formula Writing Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Naming Balancing And Formula Writing Answer Key
  - Highlighting and Note-Taking Naming Balancing And Formula Writing Answer Key
  - Interactive Elements Naming Balancing And Formula Writing Answer Key

- 8. Staying Engaged with Naming Balancing And Formula Writing Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Naming Balancing And Formula Writing Answer Key
- 9. Balancing eBooks and Physical Books Naming Balancing And Formula Writing Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Naming Balancing And Formula Writing Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Naming Balancing And Formula Writing Answer Key
  - Setting Reading Goals Naming Balancing And Formula Writing Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Naming Balancing And Formula Writing Answer Key
  - Fact-Checking eBook Content of Naming Balancing And Formula Writing Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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